## The Photoreaction of $\beta$ -Acyl- and $\beta$ -Aroyl-acrylic Acids

## Noboru Sugiyama, Hiroshi Kataoka and Choji Kashima

Department of Chemistry, Tokyo Kyoiku University, Otsuka, Tokyo

## and Kazutoshi Yamada

Departmet of Synthetic Chemistry, Chiba University, Yayoi-cho, Chiba

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During the course of an investigation of the reaction of  $\beta$ -acyl- and  $\beta$ -aroyl-acrylic acids,<sup>1,2)</sup> we have found that the methanol adduct,  $\beta$ benzoyl-α-methoxypropionic acid (1d),3) and cis- $\beta$ -benzoylacrylic acid (1b) were obtained from trans- $\beta$ -benzoylacrylic acid (1a) by irradiation in methanol.\*1 We have studied the photoreaction of la in various solvents as well as the analogous photoreactions of  $\beta$ -acyl- and  $\beta$ -aroyl-acrylic acids and their esters (2a-8a).

- $R_1=Ph R_2=H; 2) R_1=Ph, R_2=Me;$
- $R_1 = p MeC_6H_4, R_2 = H;$
- 4)  $R_1=p-MeOC_6H_4$ ,  $R_2=H$ ;
- $R_1 = p BrC_6H_4, R_2 = H;$ 5)
- 6)  $R_1=Me, R_2=H;$  7)  $R_1=R_2=Me;$
- 8)  $R_1 = t$ -Bu,  $R_2 = H$

Following three types of reaction were observed in the photoreaction listed in Table 1. a→b: trans to cis isomerization. a-c: isomerization to angelica lactones. a→d: alcohol addition to the olefinic double bond.

These reactions occurred complicatedly. For example, compound la in ether exclusively afforded the cis-isomer 1b, whereas that in methanol gave two products, the cis-isomer 1b and methanol adduct 1d. On the other hand, the irradiation of the methanol solution of 3a gave rise to both 3c and 3d. A methanol solution of 6a and 8a produced only the angelica lactone type compounds 6c and 8c. The details of these photoreactions will be reported elsewhere.

Table 1. Photoreaction products of  $\beta$ -acyl-AND β-AROYL-ACRYLIC ACIDS

Compd.	Solv.	Time (hr)	Product (%)		
			b	c	d
la	no	45	12	0	0
la	MeOH	12	17	0	14
la	MeOH	120	0	0	100a)
la	<b>EtOH</b>	72	0	0	25
la	ether	45	20	0	0
2a	MeOH	17	90	0	0
3a	MeOH	120	0	23b)	77°)
4a	MeOH	1 week	18	0	82a)
5a	MeOH	72	0	4b)	88c)
6a	no	44	0	80	0
6a	MeOH	45	0	85	0
7a	no	1 week	90	0	0
7a	acetone	17	78	0	0
8a	MeOH	31	0	100	0

a) Type d products (1d and 4d) were obtained as free acids (R2=H), which were esterified with diazomethane and subjected to NMR analysis. The methyl ester of 1d was also identified by comparison with authentic sample.2)

b), c) Irradiation gave directly methyl esters and methyl ethers.

Table 2. The NMR data  $(\tau)$  of  $\alpha$  and  $\beta$  protons OF STARTING MATERIALS AND PRODUCTS

	a	b	С	d
1	2.04,3.17	2.89,3.68	_	5.55,6.50
2	2.02,3.12	3.13,3.76	_	_
3	2.04,3.17	3.76	2.68,3.86	5.52,6.59
4	2.01,3.17	2.84,3.60	_	5.55,6.63
5	2.14,3.26	_	3.77*	5.55,6.63
6	3.20	_	2.84,3.95	_
7	3.29	3,84	_	_
8	2.37,3.24	_	2.62,3.84	

 $<sup>*</sup> R_2 = Me$ 

used as a light source and pyrex tube as a filter.

N. Sugiyama, T. Gasha, H. Kataoka and C. Kashima, This Bulletin, 41, 971 (1968).
 N. Sugiyama, T. Gasha, H. Kataoka and C. Kashima, J. Chem. Soc., in press.

<sup>3)</sup> This photoaddition of methanol was reported independently by D. B. Rao et al. in Tetrahedron Letters, p. 1613 in 1968.

\*1 Taika 500 W high pressure mercury lamp was